

# Asymmetric Construction of Quaternary Carbon Centers by Sequential Conjugate Addition of Lithium Amide and in Situ Alkylation: Utility in the Synthesis of (–)-Aspidospermidine

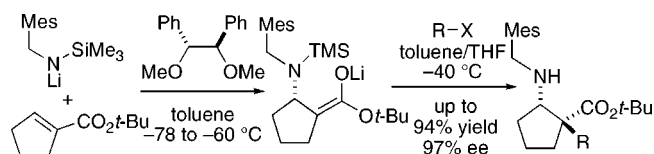
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## ABSTRACT



Chiral diether ligand-controlled asymmetric conjugate addition of a lithium amide to cyclopentenecarboxylate and subsequent in situ alkylation gave a chiral cyclopentane derivative bearing a quaternary carbon with high enantio- and diastereoselectivity. The cyclopentane derivative was converted successfully to (–)-aspidospermidine.

Synthetically useful asymmetric nitrogen–carbon bond formations that are otherwise difficult to accomplish have been achieved through conjugate addition reactions of both chiral<sup>1</sup> and achiral<sup>2</sup> lithium amides to enoates.<sup>3</sup> Since the resulting lithium enolate intermediate is a powerful nucleophile, subsequent in situ alkylation is quite promising. Thus, potentially, useful one-pot transformations to produce adjacent asymmetric carbon centers with concomitant installation

of vicinal N–C and C–C bonds could be achieved.<sup>4</sup> Of particular interest is the asymmetric construction of quaternary carbons in tandem fashion. Given our success in the development of chiral diether-mediated asymmetric conjugate additions of lithium arylmethyl- and allyltrialkylsilylamides,<sup>5</sup> we envisioned a tandem asymmetric conjugate addition–alkylation strategy for the construction of a quaternary carbon center. Herein, we describe a highly efficient enantioselective construction of a quaternary carbon center with up to 97% ee and over 98% de and the use of the product in the asymmetric total synthesis of (–)-aspidospermidine.

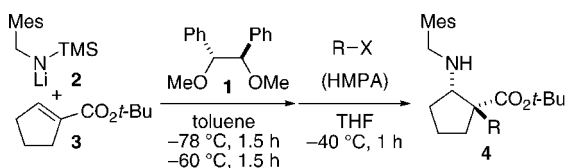
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**Table 1.** Tandem Asymmetric Conjugate Addition–Alkylation<sup>a</sup>


entry	R–X	4	yield (%)	ee (%)	de (%)
1	PhCH <sub>2</sub> Br	<b>4a</b>	90	97	>98
2	H <sub>2</sub> C=CHCH <sub>2</sub> Br	<b>4b</b>	93	95	>98
3	EtI <sup>b</sup>	<b>4c</b>	94	95	>98
4	MeI	<b>4d</b>	93	95	>98

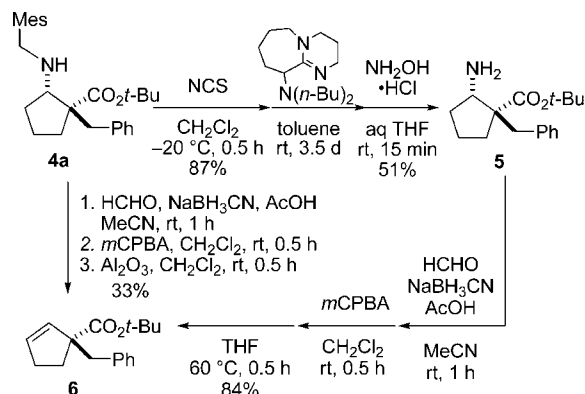
<sup>a</sup> All reactions were performed using **2** (3 equiv), **1** (3.6 equiv), and **3** (1 equiv). In the alkylation step, R–X (2 equiv) was used in entries 1, 2, and 4. <sup>b</sup> For ethylation, EtI (10 equiv) and HMPA (6 equiv) were used.

We began our studies with allylation of the lithium enolate intermediate generated by conjugate addition of lithium *N*-mesitylmethyl-*N*-TMS-amide **2** to *tert*-butyl cyclopentecarboxylate **3** in the presence of C<sub>2</sub>-symmetric chiral diether **1** in toluene at –78 °C (Table 1, entry 2).<sup>5d</sup> Treatment of a toluene solution of this lithium enolate with allyl bromide did not yield the expected allylation product **4b** (R = CH<sub>2</sub>CH=CH<sub>2</sub>), but instead yielded the conjugate addition product **4** (R = H) with 96% ee. Upon addition of THF to a toluene solution of the lithium enolate at –78 °C, the allylation process proceeded at –40 °C for 1 h, giving the desired product **4b** as nearly a single diastereomer in 92% yield. However, the enantioselectivity dropped to 90% ee. We overcame this problem by additional stirring at –60 °C for 1.5 h during the conjugate addition step (to complete the asymmetric conjugate addition, before the addition of THF, which accelerates the racemic conjugate addition reaction), giving the allylation product **4b** with 95% ee and over 98% de in 93% yield.

C-Benzoylation and C-methylation also proceeded satisfactorily to give the corresponding quaternary carbon products **4a** (R = Bn) and **4d** (R = Me) with up to 97% ee as nearly single diastereomers (entries 1 and 4). C-Ethylation with ethyl iodide required the coaddition of HMPA to give the product **4c**<sup>6</sup> (R = Et) with 95% ee in 94% yield (entry 3).

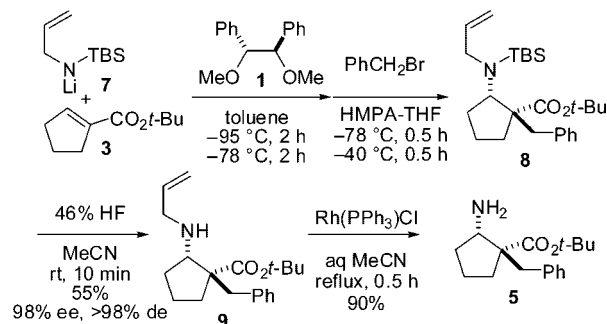
Ready conversion of the secondary amine product to the primary amine was demonstrated by successively treating **4a** with NCS for *N*-chlorination (87%); *N*-dibutylamino-DBU for elimination to the imine, and; hydroxylamine for transoximation (51%, two steps). These successive reactions gave the β-amino acid derivative **5** (Scheme 1).<sup>5d,7</sup>

Elimination of the amino function of **5** to give olefin **6** was also performed by sequential *N*-dimethylation with formalin–sodium cyanoborohydride, *m*-CPBA oxidation to

**Scheme 1.** Conversion of **4** to **5** and **6**

*N*-oxide, and thermal Cope elimination<sup>8</sup> in THF at 60 °C in 84% overall yield (Scheme 1). We also obtained olefin **6** from **4a** via *N*-methylation, *m*-CPBA oxidation, and Cope elimination<sup>9</sup> with dialuminium trioxide<sup>10</sup> in 33% overall yield.

Asymmetric conjugate addition of lithium allylamide **7**<sup>5c</sup> to **3** and in situ benzoylation gave **8**, which was then treated without purification with aq HF for protodesilylation to provide **9** (Scheme 2). Olefin isomerization of **9** with Wilkinson's catalyst in aqueous acetonitrile at reflux and simultaneous hydrolysis of the imine gave primary amine **5** in 90% yield. In this benzoylation, we obtained **9** with 98% ee in 55% yield, which is in stark contrast to 85% ee for the protonation product in the conjugate addition of **7**.<sup>5c</sup> Kinetic enantioenrichment in the benzoylation step may be responsible for this high enantioselectivity. In fact, almost complete C-allylation and C-methylation gave the corresponding products with 85% ee in reasonably high yields of 88% and 94%, respectively.

**Scheme 2.** Asymmetric Conjugate Addition of Allylamine and Benzoylation

The asymmetric total synthesis of aspidospermidine **18** has been considered to be a touchstone of the methodology for

(6) The stereochemistry of asymmetric conjugate addition has been reported in ref 5b. The relative configuration of **4c** was assigned by an NOE experiment. Ooi, T.; Miki, T.; Taniguchi, M.; Shirashi, M.; Takeuchi, M.; Maruoka, K. *Angew. Chem., Int. Ed.* **2003**, *42*, 3796–3798.

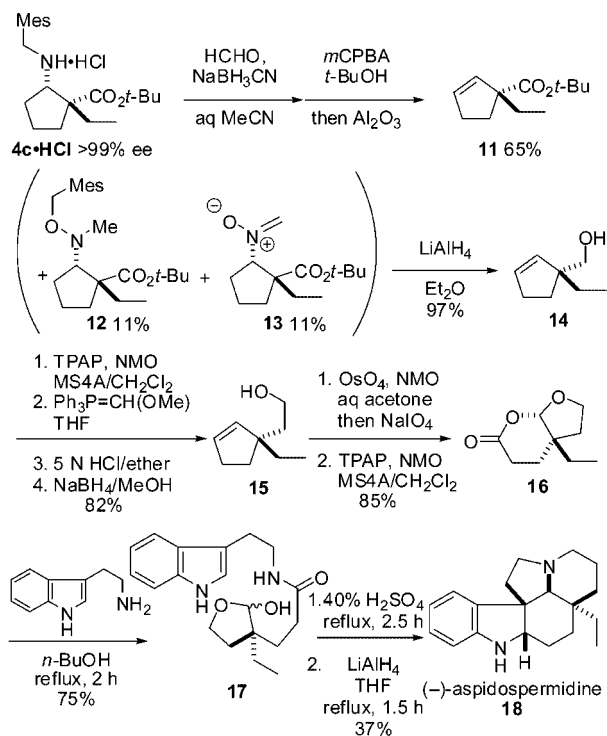
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### Scheme 3. Total Synthesis of (–)-Aspidospermidine 18



the asymmetric construction of quaternary carbons.<sup>11,12</sup> The utility of the quaternary carbon product **4** was demonstrated by the asymmetric total synthesis of (–)-**18** (Scheme 3).<sup>13</sup>

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The ee of the ethylation product **4c** (95% ee) was increased to >99% ee in 85% recovery by recrystallization of its hydrochloride from ethyl acetate. Successive methylation of the hydrochloride salt of **4c**, *N*-oxidation with *m*-CPBA, and Cope elimination with Al<sub>2</sub>O<sub>3</sub> in *tert*-butyl alcohol<sup>14</sup> gave olefin **11** in 65% yield along with **12** (11%) and **13** (11%). Lithium aluminum hydride reduction of ester **11** to alcohol **14**, TPAP oxidation to the aldehyde, Wittig olefination for one-carbon elongation, hydrolysis to the aldehyde, and sodium borohydride reduction gave alcohol **15** in 79% yield (five steps from **11**). These steps should be carefully carried out because the products all have low boiling points and can easily be distilled off. Oxidative cleavage of olefin **15** with osmium tetroxide and sodium metaperiodate yielded a lactol, which was then oxidized to  $\delta$ -lactone **16**, bearing three differently oxidized oxygen functionalities<sup>15</sup> in 85% yield. Amide formation with lactone **16** and tryptamine in *n*-butyl alcohol at reflux gave **17** in 75% yield. The remaining transformations were a modification of the Harley–Mason’s protocol. Sulfuric acid treatment of **17** induced Pictet–Spengler cyclization. Simultaneous rearrangement occurred at reflux for 2.5 h. Lithium aluminum hydride reduction in THF at reflux for 1.5 h completed the synthesis of (–)-aspidospermidine **18** in 37% yield.<sup>16</sup>

In summary, we have developed an efficient method for constructing a chiral quaternary carbon by tandem asymmetric conjugate addition of a lithium amide to an enoate and its subsequent *in situ* alkylation. Total synthesis of (–)-aspidospermidine was successfully demonstrated as a touchstone of the strategic application of the asymmetric conjugate amination–alkylation protocol.

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**Supporting Information Available:** Experimental procedures and characterization data of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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